

CO₂ Capture & Treatment in Craft Brewing

Introduction

Carbon dioxide (CO₂), is both a by-product and a necessary component of the beer brewing process. Many large capacity breweries have capture and treatment systems in place due to their large production rates. Currently, no such system is designed for small capacity breweries.

Project Statement:

Design a system for TUNS Brewery that can collect, purify and recycle the CO₂ for use in the brewhouse.

Constraints:



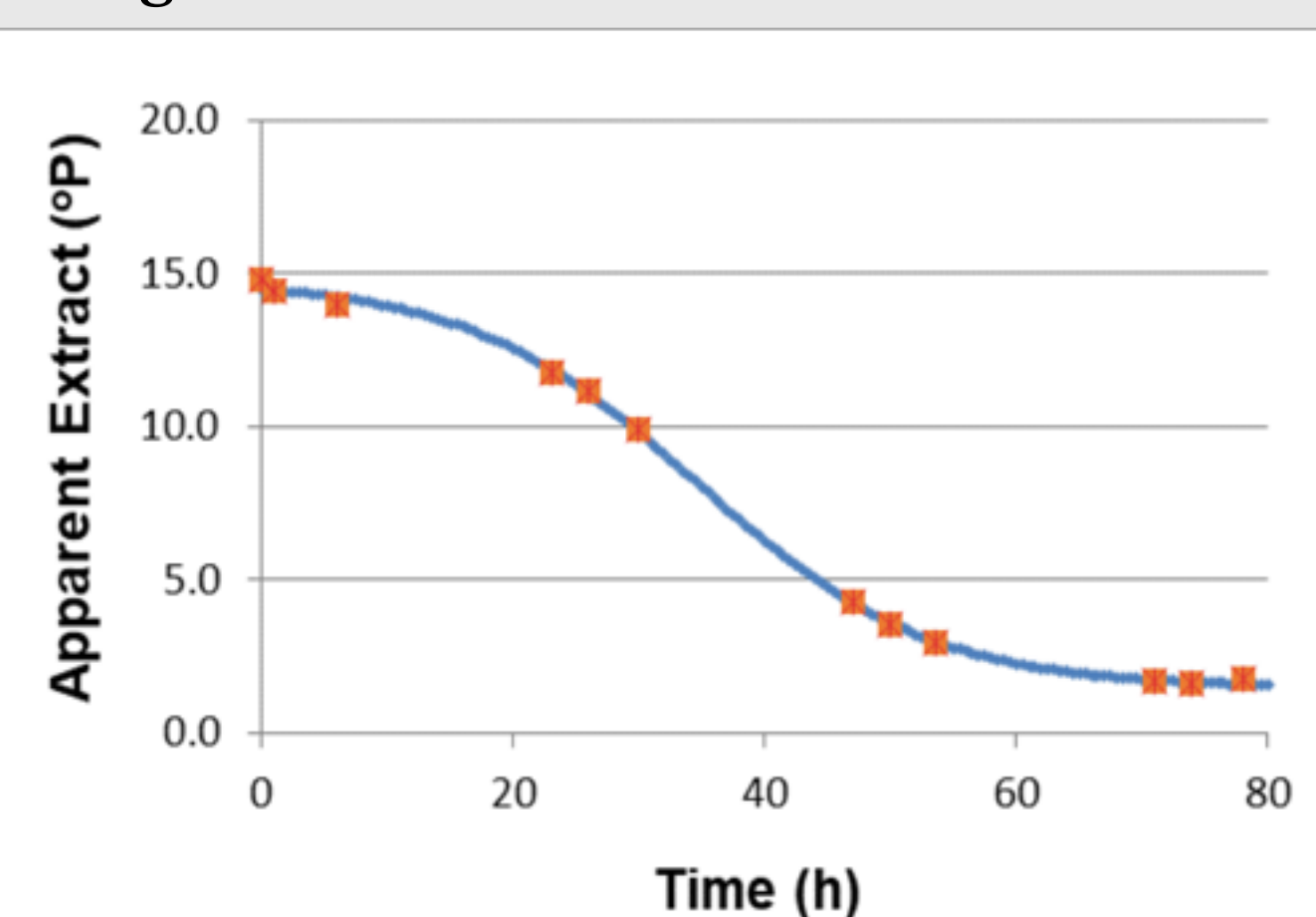
- 1 Cost
- 2 Size
- 3 Food Grade
- 4 Complexity

Fermentation Process

- Wort sugar (glucose) is converted into ethanol and CO₂ as shown below:



- Collection begins at hour 35 and ends at hour 70



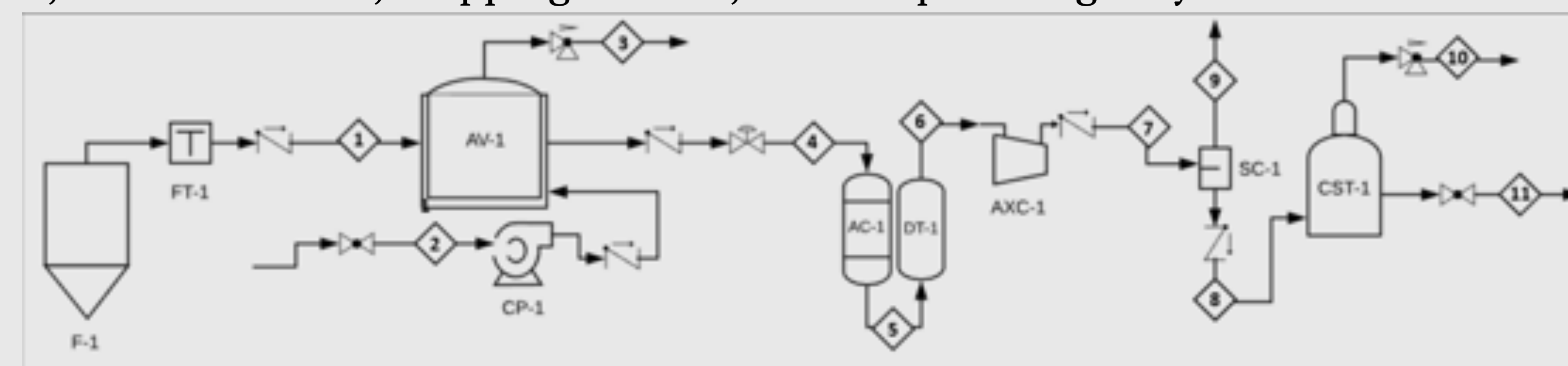
Fermentation model made using logistic model as detailed by Speers et al (2005)

Safety Considerations

- Pressurized System
- CO₂ Asphyxiation
- Compressed gas
- Food grade

Design Overview

The TUNS Brewery CO₂ collection and purification system is comprised of a foam trap, accumulator vessel, adsorbent bed, desiccant tower, stripping column, and compressed gas cylinder.



1. Collection

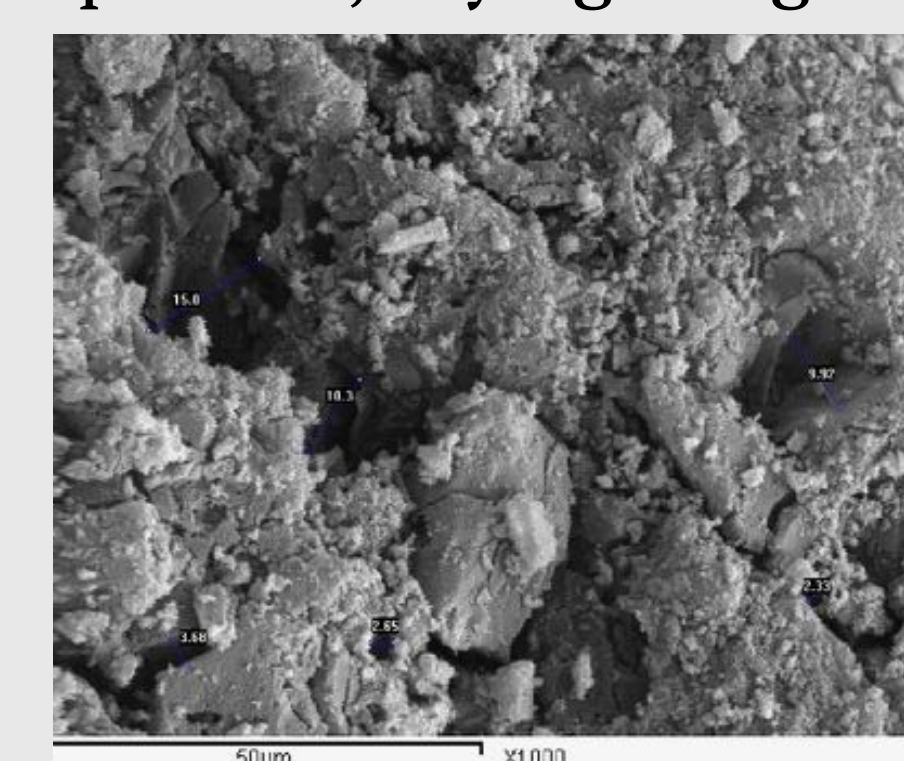
- Effluent gas will exit the fermenter and flow through a foam trap into a bladder accumulator vessel
- The bladder prevents water (the pressure medium) from contacting the effluent gas
- The accumulator vessel will pressurize the effluent gas to 5 bar



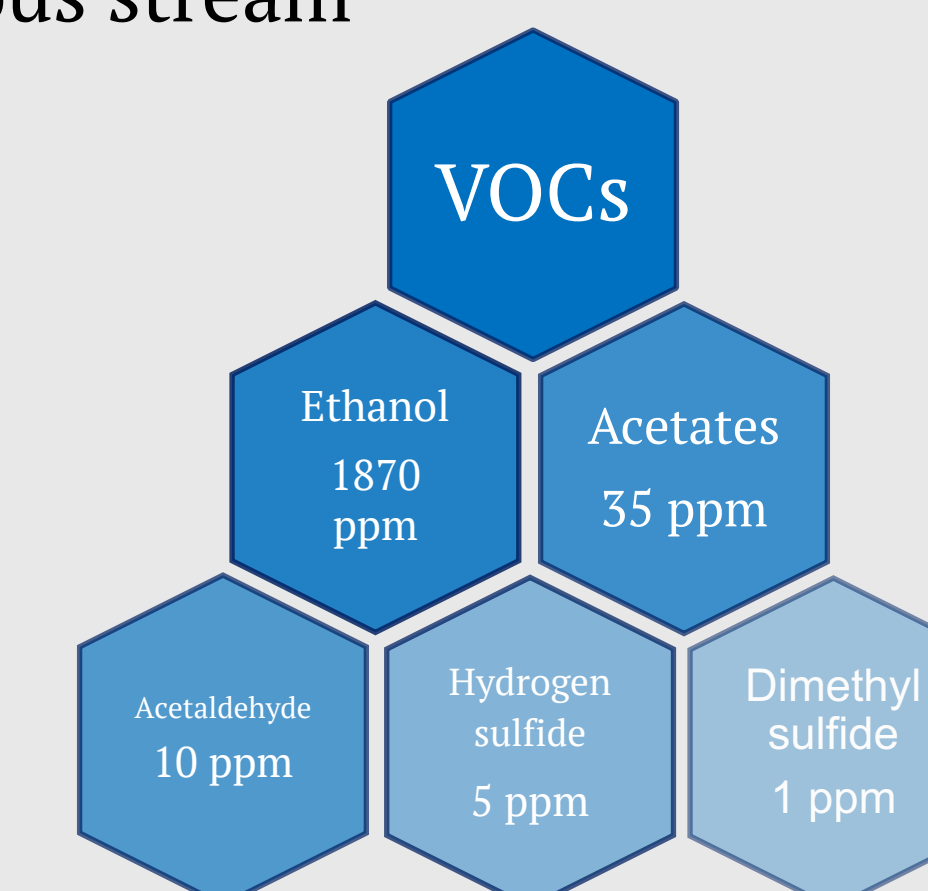
Bladder accumulator vessel from Parker Hannifin (2016)

2. VOCs and H₂O Removal

- The activated carbon adsorption bed targets the volatile organic compounds found in the effluent gas
- BASF F-200 activated alumina adsorbs any water present, drying the gaseous stream



Activated carbon micrograph taken with an electron microscope (Abdel-Wahhab, 2015)



4. Storage

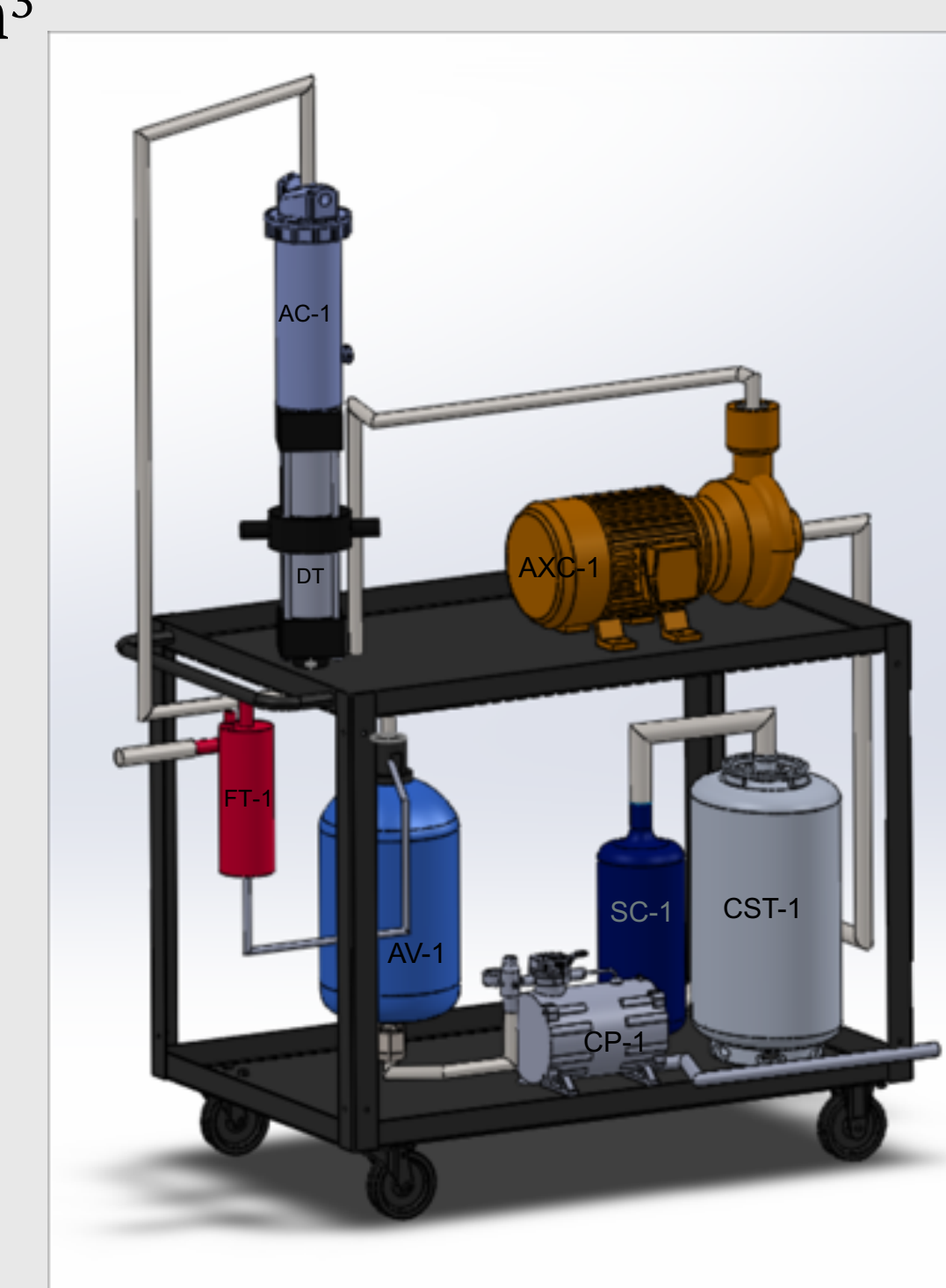
- The purified CO₂ will be stored as a compressed gas in a gas cylinder
- The storage pressure is 50 bar



3D SolidWorks Rendering

Size constraint: 2.14 m³

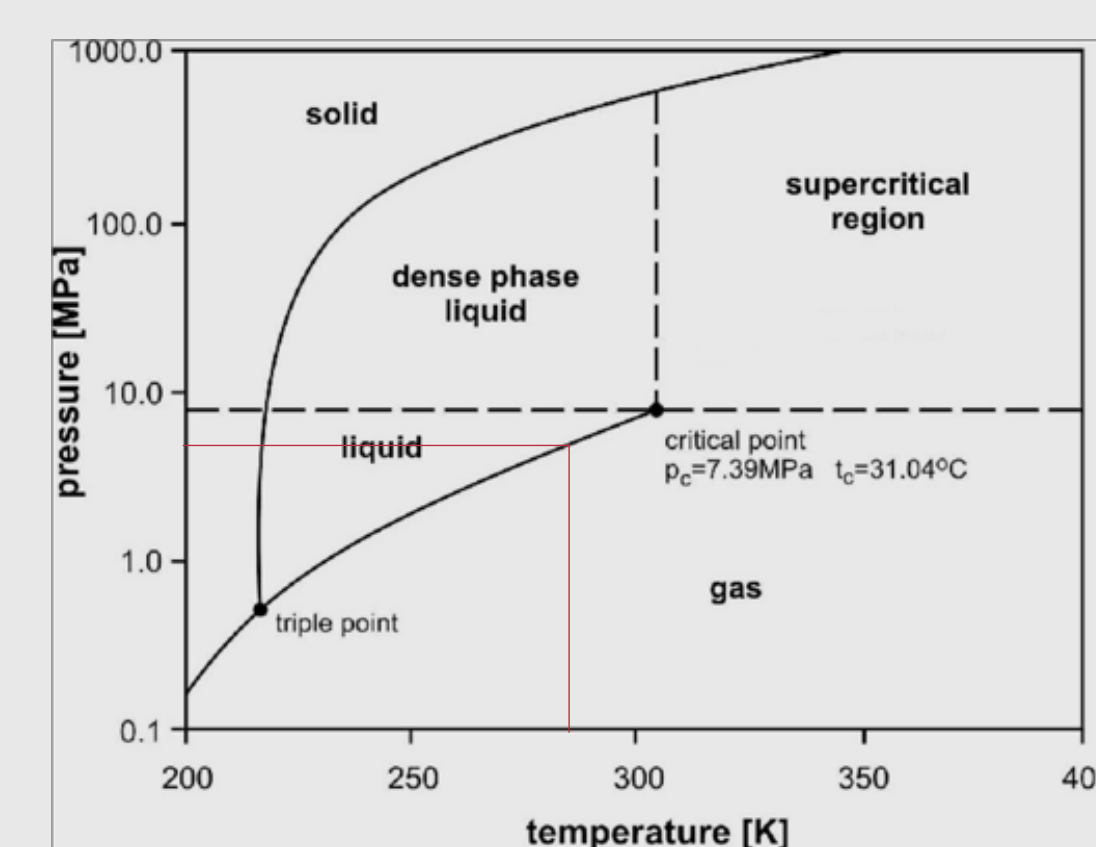
- Height: 1.52 m
- Width: 0.84 m
- Length: 1.68 m



System modeled in SolidWorks

3. O₂ Removal

- O₂ must be removed from beverage grade CO₂ to prevent degradation of taste in beer
- The CO₂ stream is compressed to 55 bar then cooled to room temperature for liquefaction
- Next, O₂ is stripped from the CO₂ using N₂



CO₂ Phase Diagram indicating operating conditions (Witkowski, 2014)

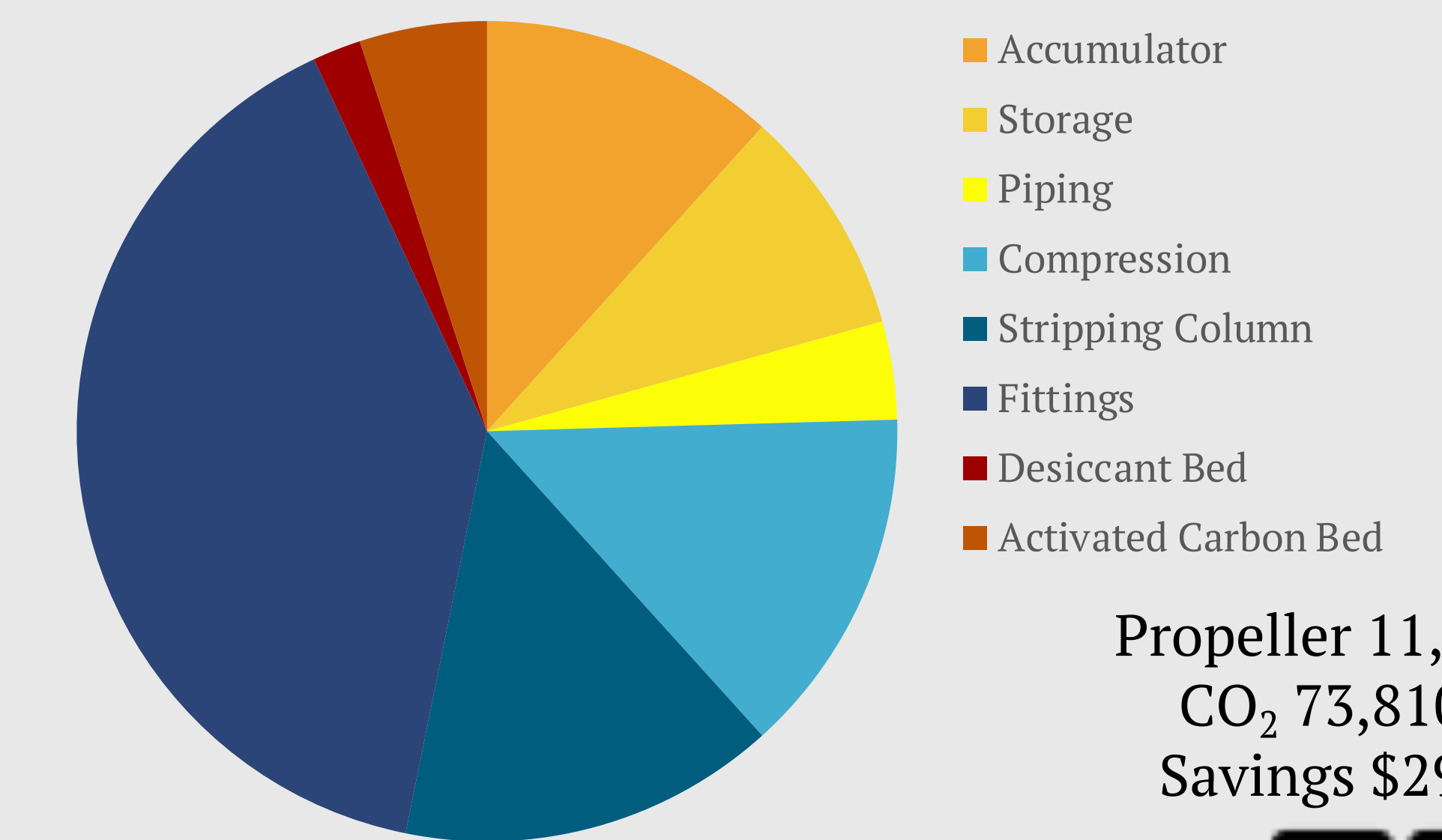
ISBT Guidelines

Contaminant	Max allowed
Purity	99.9% min
Moisture	20 ppm max
Oxygen	30 ppm max
Total volatile hydrocarbons	50 ppm max
Acetaldehyde	0.2 ppm max
Aromatic hydrocarbon	20 ppb max
Total sulfur content	0.1 ppm max
Sulfur dioxide	1 ppm max

Guidelines for beverage grade CO₂ (Brewers Association Draught Beer Quality Manual)

Economics

The cost estimate for the TUNS treatment system is \$4600



Propeller 11,000 hL/yr
CO₂ 73,810 lbs/yr
Savings \$29,000/yr

Cost breakdown for TUNS system

Good Robot 1070 hL/yr
CO₂ 7175 lbs/yr
Savings \$2800/yr

TUNS 48 hL/yr
CO₂ 322 lbs/yr
Savings \$500/yr



Conclusion

- Designed system can purify and recycle CO₂ from TUNS fermentation
- TUNS scale is very small compared to many Nova Scotian craft breweries resulting in small CO₂ collection

Future Work & Recommendations

- The system could be economically feasible if successfully scaled to larger craft brewery (>10,000 hL/yr)
- A more accurate assessment of cost scaling for larger system should be performed
- Prototype building and testing for feasibility would need to be completed before pursuing a larger system

References & Acknowledgments

Abdel-Wahhab, M., Aziza, H., Abdel-Azium, S., El-Kady, A., Abd El-Moneim, O. (2015). Effectiveness of activated carbon and Egyptian montmorillonite in the protection against deoxynivalenol-induced cytotoxicity and genotoxicity in rats. *Food and Chemical Toxicology*. 83, 174-182.

Parker Hannifin (2016).

Speers, R. A., Rogers, P., & Smith, B. (2005). Non-linear modelling of industrial brewing fermentations. *Journal of the Institute of Brewing*, 109(3), 229-235.

Witkowski, A., Majkut, M., & Rulik, S. (2014). Analysis of pipeline transportation systems for carbon dioxide sequestration. *Archives of Thermodynamics*, 35(1), 117-140.

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